

An Investigation Into The Mechanisms Underlying Enhanced Biosulphidogenesis In *Bacillus* Spp.

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Abstract: Anthropogenic activities like mining, processes of metallurgy and other chemical industries lead to the discharge of a high amount of sulphate into the environment that causes serious problems to human health. This paper illustrates the employment of thermophilic sulphate reducing bacteria for biosulphidogenesis. Two different species have been isolated from hot water spring of Vajreshwari and Ganeshpuri, Thane, Maharashtra, INDIA. The mechanism involved in biosulphidogenesis includes production of specific protein as well as liberation of some extracellular polymeric compound (EPS) e.g. proteins, carbohydrate, acids etc. that are produced during the microbial cell metabolism. These compounds play an important role in the faster reduction of sulphate and decrease in production rate of sulphide. The isolate was found to be of genus *Bacillus* and type strain was found to be *subtilis* Zankar and *licheniformis* Sonali. The strain sequence were deposited in NCBI database with accession number KJ939324 and KJ939325 respectively. The result highlights the potential use of these organism in biosulphidogenesis.

Keywords: Sulphate reducing bacteria, Protein Expression, Extracellular polymeric Substances, FTIR.

I. Introduction

Thermophilic Bacteria are ubiquitous and highly diverse. They can survive in all sorts of inhospitable environments. Studies in the last two decades have revealed that 99% of bacteria present in the environment are still unexplored or overlooked in laboratory cultivation and hence remain obscure for their ecological functions and unexploited for biotechnological applications.

One of the best application of thermophiles is reduction of sulphate. Sulphate Reducing Bacteria (SRB) are currently subjected to extensive genomic studies, to understand their basic biochemical mechanisms, which would help in the development of novel techniques for the analyses of their role in environment. SRB uses different sort of mechanism for reduction of sulphate, which includes production of specific enzymes, extracellular polymeric substances (EPS), etc.

The expression of metal-binding proteins or peptides in microorganisms in order to enhance sulphate accumulation and/or tolerance has great potential. Several different peptides and proteins have been explored. The mechanisms involved in bacterial metal resistance result from either the active efflux pumping of the toxic metal out of the cell, or the enzymatic detoxification (generally redox chemistry) converting a toxic ion into a less toxic or less available metal ion.

Another mechanism is EPS, which are a complex high-molecular-weight mixture of polymers excreted by microorganisms, produced from cell lysis and adsorbed organic matter from wastewater. Their characteristics (e.g., adsorption abilities, biodegradability and hydrophilicity/hydrophobicity) and the contents of the main components (e.g., carbohydrates, proteins, humic substances and nucleic acids) in EPS are found to be crucially affecting the properties of microbial aggregates, such as mass transfer, surface characteristics, adsorption ability, stability, the formation of microbial aggregates etc.

So the objectives of this study was to isolate thermophilic sulphate reducing bacteria, identify them and to investigate the mechanism of sulphate reduction present in the isolated species.

II. Methods And Materials

Sampling:

Water samples were collected from seven different hot springs (from Vajreshwari & Ganeshpuri, Thane) Mumbai. Surface water samples were taken from the Hot Springs using a grab sampler. A 500-ml plastic cup attached to a 2-m pole was dipped into the water twice to rinse it. The sample was then transferred to a clean, new, polyethylene container with a snap-on lid. The temperature of the sample was taken with a laboratory thermometer and recorded. All samples were taken on the same day to prevent discrepancies due to sample date. Samples were kept cool during transport to the laboratory and processed within 12 h of collection (Vieille C., 1996). Spring water used for inorganic analyses was immediately fixed with 3 ml of a 20% (wt/vol) zinc acetate solution for sulfate analysis & with the same volume of a 90mM zinc acetate solution for determination of the sulfide concentration. In situ spring water used in the test for sulfide production by TSRP was kept in a sterile plastic bottle (1,000ml).

Media & growth Conditions

Medium D described by Postgate (1969) was used for routine stock maintenance and all enrichment culture studies. Medium D contained (g/ Lit.: Na Pyruvate - 3.5g , MgCl₂.6H₂O - 1.6g , NH₄Cl - 1g , Yeast Extract - 1g , KH₂PO₄ - 0.5g , CaCl₂.2H₂O - 0.1g , FeSO₄.7H₂O - 0.04g , pH 7.5 ± 0.2 (25°C). Black colonies were isolated from anaerobic roll tubes (Hungate, 1969) containing Medium D and 4% (w/v) purified agar. Stock cultures of all strains were prepared from single isolated colonies that proliferated on transfer in Medium D. All stock cultures were incubated at 50°C. Cultures were routinely checked for contamination by the detection of growth in the complex medium, TYEG, described for growth of saccharolytic thermophiles (Zeikus *et al.*, 1979).

Characterization and identification of the isolates :

I) Morphological Studies :

Morphological properties were investigated by using 18 hour old bacterial cultures. These included the wet mount preparations using light microscope & Gram staining to confirm Gram reaction. Motility was determined by hanging drop method.

II) Biochemical Tests :

Two thermophilic isolate was identified by the use of conventional methods for the presumptive identification of physiological and biochemical tests. These tests were; Gram reaction, catalase production, hydrolysis of protein, starch and lipid, and acid production from sugar (**Campos, et al**,1995). The species was reconfirmed by using automated Biomerieux Vitek 2 System (Nucleus Diagnostic Centre, Kalyan).

Biosulphidogenesis:

a) Sulfate reduction rate (SRR) :

Measurement of the reduction of sulphate into sulphide was performed via the exact procedures detailed by Ingvorsen *et al.* (1981) for determination of bacterial sulphate reduction in anoxic water and sediment samples. The rate of sulphate reduction was determined from points taken during a 10 hr time course. (All assays were done in triplet.)

b) Sulfide production rate :

Dissolved sulfide concentrations were analysed by using Iodometric method (APHA) after in situ fixation with a zinc acetate solution. Concentration of sulfide was determined both before & after process of sulphate reduction to estimate actual amount produced by biosulphidogenesis. (All assays were done in triplet.)

Mechanisms involved in biosulphidogenesis :

a) Isolation of Proteins :

The bacterial cell of both the strains were grown at 45 C for 24 hrs. and the cells were harvested by centrifugation at 13,000 rpm for 10 mins. Later the cell pellet was washed with phosphate buffer (pH 7.0) to remove the traces of remaining media and again centrifuged at 10,000 rpm for 10 mins. The supernatant was discarded. The cell pellet obtained was mixed with 1ml of 2X sample buffer (0.5% SDS, 25% beta mercaptoethanol, 0.03% bromophenol blue, 2.5% Glycerol, 15mM trisHCl (pH 6.8)). The samples were vortexed and incubated in a boiling water bath for half an hour. The samples were used directly for SDS-PAGE analysis (Maiti *et al.*, 2009). The same method of isolation was performed with and without the treatment of sulphate (control and Test).

The most probable protein matching according to molecular weight was carried out by insilico method using by <http://web.expasy.org/Database>.

b) Isolation of EPS :

Six day-culture broths (250ml nutrient broth containing test organisms incubated at 50°C at 80rpm) were centrifuged (6000 ×g, 30mins, 4°C). the EPS were isolated either (i) from bacterial pellets or (ii) from supernatants.

- i. The cell pellets from the culture were suspended in 10ml (20g /liter) NaCl in water and were centrifuged (6000 ×g, 30mins, 4°C). the resulting pellets were suspended in 10ml (10g /liter) NaCl in water and were dialyzed against water for 48 hours. The dialyzed samples were centrifuged (6000 ×g, 30mins, 4°C), and stored at 4°C.
- ii. The supernatants were filtered through Whatman Filter paper no 40. After the addition of cold ethanol (ethanol/filtrate ratio 2:1), the solutions were kept overnight at 4°C. The EPS precipitates were recovered by centrifugation (6000 ×g, 30mins, 4°C).

The pellets were suspended in water and stored at 4°C. The protein contents of EPS was determined by Barfoeds method and total neutral- carbohydrate content was determined by phenol-sulphuric acid method and were confirmed by FTIR analysis. The same method of isolation was performed with and without the treatment of sulphate (control and Test) (Fabienne Francois, *et. al.*2011).

FTIR Analysis for EPS

Samples for FTIR can be prepared in a number of ways. For liquid samples, the easiest is to place one drop of sample between two plates of KBr (salt). Salt is transparent to infrared light. The drop forms a thin film between the Plates which was analysed by FTIR. The plates were scanned from 400 to 4000 wavenumber. The functional group from each spectra was determined using co-relation chart.

Strain identification:

It was done by 16s rRNA Analysis and the isolated colonies were sequenced for its conserved sequences and analysed for partial 16s rRNA by geneOmbio, Pune, Maharashtra.

The predicted 16S rRNA sequences from this study were compared with 16S rRNA sequences in a BLAST database constructed from sequences downloaded from the Ribosomal Database Project (release 8.1; <http://rdp8.cme.msu.edu>). Comparisons were made using the program BLAST (<ftp://ftp.ncbi.nih.gov/BLAST/executables/LATEST/>) and a FASTA-formatted file containing the predicted 16S rRNA sequences.

III. Results

Characterization of in situ sulphidogenesis:

Microbial sulphate reduction at high temperatures was studied at several sites in Vajreshwari & Ganeshpuri hot springs. Enrichment cultures were initiated with Medium D. After incubation of all enrichments for 6 d at *in situ* temperature, the cultures formed a dense black precipitate. Repeated transfer of enrichments revealed small rod-shaped sulphate reducing bacteria in all cultures. All strains appeared morphologically identical with one being mucoid colony (SZP 1) and one with smooth colony (SZP 2) when cultured on Medium D at 41°C. Stock cultures of both strain were maintained on Medium D and transferred monthly.

Cellular properties:

Cells appeared as very tiny straight rods. Motility was not observed. Exponential phase cells stained Gram-positive. Other biochemical properties (As per Berget,s manual) are described in Table 1. Species was again confirmed by using automated system of Biomerieux System (Done at Nucleus diagnostic Centre, Kalyan) stated as in Table 2 and 3.

Table 1: Biochemical Properties of TSRP (According to Bergey's manual)

Characteristics	Strain SZP 1	Strain SZP 2
Gram Nature	Positive	Positive
Shape	Bacillus	Bacillus
Motility	M	M
Temperature°C– Optimum	41	45
pH - Optimum	6.5	7.0
NaCl % - Optimum	1.5	1.4
Oxidase	+	+
Catalase	+	+
Casein	+	+
Starch	+	+
D-Glucose	+	+
D-Fructose	+	-
Maltose	-	+
Mannose	-	-
Trehalose	-	-
Cellobiose	-	-
Sucrose	-	-
Melibiose	+	+
Lactose	+	+
Xylose	-	+
Galactose	-	-
Nitrate	-	-
Citrate	+	+

Key : Growth : + ; No Growth : - ; M : Motile

Table 2: Biochemical Tests (ByBiomerieuxVitek 2 System) For SZP 1

Biochemical Details																	
2	AMY	+	4	PIPLC	+	5	dXYL	-	8	ADH1	+	9	BGAL	-	11	AGLU	+
13	APPA	-	14	CDEX	-	15	AspA	+	16	BGAR	-	17	AMAN	-	19	PHOS	+
20	LeuA	-	23	ProA	-	24	BGURr	+	25	AGAL	-	26	PyrA	-	27	BGUR	+
28	AlaA	+	29	TyrA	+	30	dSOR	-	31	URE	+	32	POLYB	+	37	dGAL	-
38	dRIB	-	39	ILATk	+	42	LAC	+	44	NAG	+	45	dMAL	-	46	BACI	-
47	NOVO	-	50	NC6.5	+	52	dMAN	-	53	dMNE	+	54	MBdG	+	56	PUL	-
57	dRAF	-	58	0129R	-	59	SAL	+	60	SAC	-	62	dNTRE	-	63	ADH2S	+
64	OPTO	-															

Table 3: Biochemical Tests (ByBiomerieuxVitek 2 System) For SZP 2

Biochemical Details																	
2	AMY	--	4	PIPLC	+	5	dXYL	-	8	ADH1	-	9	BGAL	+	11	AGLU	+
13	APPA	+	14	CDEX	+	15	AspA	-	16	BGAR	+	17	AMAN	+	19	PHOS	-
20	LeuA	-	23	ProA	-	24	BGURr	-	25	AGAL	-	26	PyrA	-	27	BGUR	-
28	AlaA	-	29	TyrA	-	30	dSOR	-	31	URE	+	32	POLYB	-	37	dGAL	-
38	dRIB	-	39	ILATk	+	42	LAC	-	44	NAG	+	45	dMAL	+	46	BACI	+
47	NOVO	-	50	NC6.5	-	52	dMAN	-	53	dMNE	+	54	MBdG	-	56	PUL	-
57	dRAF	-	58	0129R	-	59	SAL	-	60	SAC	-	62	dNTRE	+	63	ADH2S	+
64	OPTO	-															

Biosulphidogenesis:

The MIC for both the strains were found to be above 1000 ppm. The reduction rate for standard Sulphate was found to be 100% in 14 hrs. for SZP 1 and 90% for SZP 2 in 14 hrs. with no sulfide production, Table 4.

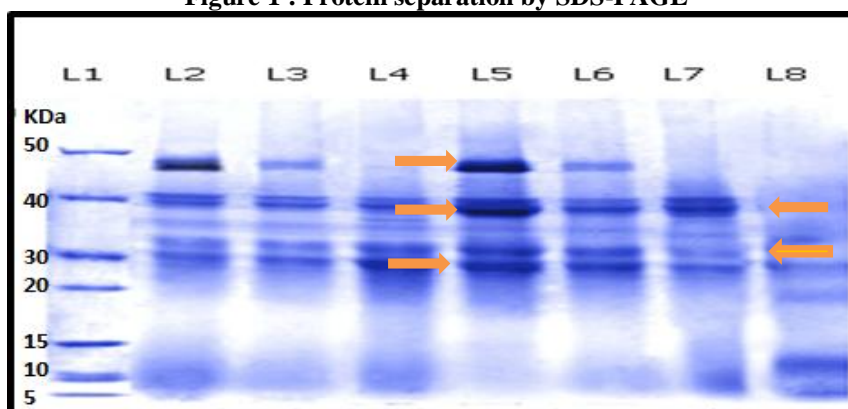
Table 4: Initial & final concentration of sulphate&sulphide before & after the treatment

Strain	Sulphate concentration (mg/L)			Sulphate Reduction rate (%)	Sulphide concentration (mg/L)		Sulphide Production rate (%)
	Initial	Final	Reduction Time (hrs.)		Initial	Final	
SZP 1	1000	0	14	100	--	--	0
SZP 2	1000	900	14	90	--	--	0

Protein Isolation ;

The gel showed that in presence of sulphate some of the protein/enzyme for both the strains are over expressed. These protein were identified by insilico method and most probable protein matching with the observed results are shown in figure.

Figure 1 : Protein separation by SDS-PAGE



Key: Lane 1 : Protein Marker, Lane 6: *Bacillus subtilis* Zankar Untreated, Lane 5 *Bacillus subtilis* Zankar treated, Lane 7 : *Bacillus licheniformis* Sonali treated, Lane 8: *Bacillus licheniformis* Sonali untreated. According to molecular weight found, the following protein can be probably expressed during Biosulphidogenesis.

***Bacillus subtilis* :**

- 1) Cytochrome c-oxidoreductase (28,162 Kda)
- 2) sulphatepermease (36,836 kDa)
- 3) sulfate adenylyltransferase (42884 KDa)

***Bacillus licheniformis* : adenylylsulfate kinase (22456 KDa)**

EPS Production:

The EPS was extracted from both the organisms and colorimetric quantification of the proteins and carbohydrates was performed. The results indicated the high amount of both proteins and carbohydrate (Table 5) in case of test which was further confirmed by FTIR.

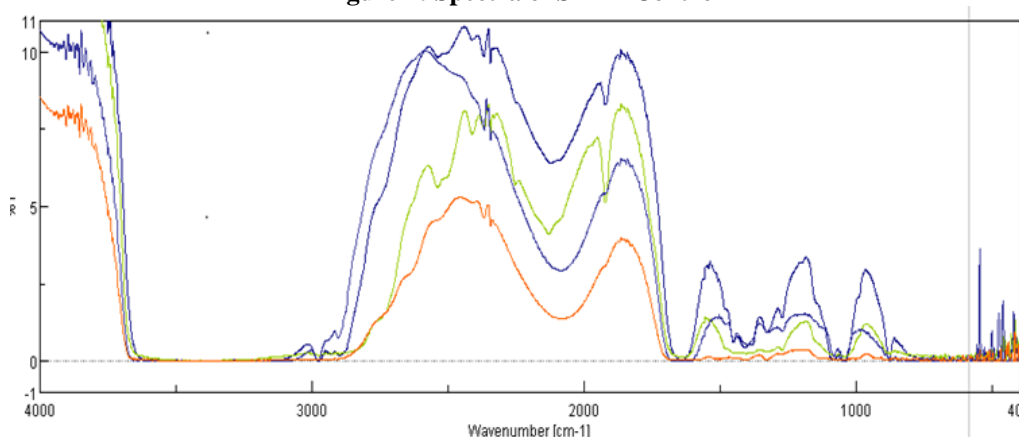
Table 5: Amount of carbohydrate and protein present in EPS for Control and Test (in treated with sulphate) for strain SZP 1 and SZP 2.

Strain	Carbohydrate Content ppm		Protein Content ppm	
	Control	Test	Control	Test
SZP 1	80	89	56	93
SZP 2	68	75	79	84

Fourier Transform – Infrared Spectroscopy Analysis:

The characterization of EPS by (FT-IR) spectroscopy showed the typical absorption band for carboxylic acid (a broad O—H elongation band in the range 2100 – 3500 cm⁻¹) as well as bands for aldehydes and ketones (a broad H—C=O, C—H stretch in the range 1500-2000cm⁻¹) and a band for 1° amines (a N—H stretch in the range 750-800 cm⁻¹). Thus, the relative measure of intensities confirmed the presence of proteins and carbohydrate in agreement with the colorimetric quantification.

Figure 2: Spectra of SZP 1 Control



Key: Orange Line: SZP 1 Control ; Blue Line – SZP 1 Treated ; Green line - SZP 2 Control ; Purple Line – SZP 2 Treated

16s rRNA Analysis :

The partial sequence analysis of conserved sites of both the strain gave the following sequences.

SZP 1

TTGGCCCCTGCTCTTATCCTTGCCAGTCGAGCGGACAGATGGGAGCTTGCTCCCTGATGTTAGCGG
 CGGACGGGTGAGTAACACGTGGGTAACCTGCCTGTAAGACTGGGATAACTCCGGGAAACCGGGG
 TAATACCGGATGCTGTTTGAACCGCATGGTTCAAACATAAAAGGTGGCTTCGGCTACCACTT
 ACAGATGGACCCGCGGCATTAGCTAGTTGGTGAGGTAATGGCTCACAAGGCAACGATGCGTAG
 CCGACCTGAGAGGGTGATCGGCCACACTGGGACTGAGACACGGCCAGACTCCTACGGGAGGCAG
 CAGTAGGGAATCTCCGCAATGGACGAAAGTCTGACGGAGCAACGCCGCGTGAGTGATGAAGTTTT
 CGGATCGTAAAGCTCTGTTGTTAGGGAAGAACAAGTACCGTTCGAATAGGGCGGTACCTTGACGG
 TACCTAACAGAAAGCCACGGCTAACTACGTGCCAGCAGCCGCGGTAATACGT

SZP 2

GGGTGAGTAACACGTGGGTAACCTGCCTGTAAGACTGGGATAACTCCGGGAAACCGGGGCTAATA
 CCGGATGCTTGATTGAACCGCATGGTTCAATCATAAAAGGTGGCTTTTAGCTACCACTTGCAGATG
 GACCCGCGGCGCATTAGCTAGTTGGTGAGGTAACGGCTACCAAGGCGACGATGCGTAGCCGACC
 TGAGAGGGTGATCGGCCACACTGGGACTGAGACACGGCCAGACTCCTACGGGAGGCAGCAGTAG
 GGAATCTTCCGCAATGGACGAAAGTCTGACGGAGCAACGCCGCGTGAGTGATGAAGGTTTTCGGA
 TCGTAAAACTCTGTTGTTAGGGAAGAACAAGTACCGTTCGAATAGGGCGGTACCTTGACGGTACCT
 AACCAGAAAGCCACGGCTAACTACGTGCCAGCAGCCGCGGTAATACGTAGGTGGCAAGCGTTGTC
 CGGAATTATTGGGCGTAAAGCGCGCGCAGGCGGTTTCTTAAGTCTGATGTGAAAGCCCCCGGCTCA
 ACCGGGAGGGTCATTGGAAACTGGGGAACCTGAGTGCAGAAGAGGAGAGTGGAATTCCACGTGT

AGCGGTGAAATGCGTAGAGATGTGGAGGAACACCAGTGGCGAAGGCGACTCTCTGGTCTTGTAAC
TGACGTTGAAGGCGGGGCTAAAGGGA

The sequences were aligned for comparison with the existing databases by using in silico tool BLAST. The isolate was found to be of genus *Bacillus* and type strain was found to be *subtilis Zankar and licheniformis Sonali*. The strain sequence were deposited in NCBI database with accession number KJ939324 and KJ939325 respectively.

IV. Discussion

The present study extends the known niche for bacterial assimilatory sulphate reduction in nature to extreme (> 60°C) thermal environments. Bacterial sulphate reduction at high temperatures appears widespread in Vajreshwari & Ganeshpuri hot springs and was associated with one genus with two different colony characteristics. Biochemical test (Done by Manual and automated method) and also 16s rRNA analysis showed that species found to be of Genus *Bacillus* & species as *subtilis Zankar and licheniformis Sonali*. Both the strains showed tolerance level of sulphate upto 1000 ppm. Thermophilic sulphate-reducing bacteria were found in environments where organic matter was being vigorously decomposed. Biosulphidogenesis appeared most active in the sulphate-depleted thermal ecosystem, where microbial sulfate reduction occurred.

Both strains showed 100 % and 90% reduction of standard sulphate in 14hr with no production of sulphide production. Experiments by J. Suschka and L. Przywara shows that 90 % sulphate reduction was achieved after 60hrs. by using *Desulfovibrio* or *Desulfobacterium* with sulfide production as high as 150 mg/L.

The specified proteins were expressed during the biosulphidogenesis showed that various operons are responsible for this process. The chemical analysis of EPS and the FT-IR spectroscopy analysis confirmed the presence of proteins and carbohydrates. Further study needs to be performed for identification and characterization of the same and for the use of these organisms for bioremediation of heavy metals. As, the bioremediation of inorganic material is important for health of the people & for monitoring environment, in situ sulphidogenesis by both *Bacillus* species is one of the best ways for the treatment of industrial effluent containing high levels of sulphate.

References

- [1]. Cloetete., Gerber a. and Maritz I. V., A first order inventory of water use and effluent production by SA industrial, mining and electricity generation sectors, Arcusgibb (pty) Ltd, water resource division (water –north), pretoria, wrc report no. 1547/1/10 (2010)
- [2]. Krishnanandan Vivek, Srikantaswamy S, Assessment of impacts by Industries on sediments of Kabini river around Nanjangud Industrial area, Karnataka, India, International Journal of Scientific and Engineering Research, 4(11)(2013)
- [3]. Vemula M., Ambavaram V., Kalluru G. and Gajulapalle M., Tollamadugu N. An Overview on Research Trends in Remediation of Chromium, Research Journal of Recent Sciences, 2(1), 71-83, January (2013)
- [4]. Anna H. Kaksonen, Jason J. Plumb, Wendy J. Robertson, Stefan Spring, Peter Schumann, Peter D. Franzmann and Jaakko A. Puhakka, Novel Thermophilic Sulfate-Reducing Bacteria From A Geothermally Active Underground Mine In Japan. Appl Environ Microbiol., May, 72(5), 3759– 3762 (2006)
- [5]. Carine Audiffren, Jean-Luc Cayol, Catherine Joulain, Laurence Casalot, Pierre Thomas, Jean-Louis Garcia and Bernard Ollivier, *Desulfonauticus submarinus* gen. nov., sp. nov., a novel sulfate-reducing bacterium isolated from a deep-sea hydrothermal vent, International Journal of Systematic and Evolutionary Microbiology, (5), 1585-1590 (2003)
- [6]. Fabienne Francois, Carine Lombard, Jean-Michel Guigner, et. al., Isolation and characterization of Environmental bacteria capable of extracellular biosorption of mercury, Applied and Environmental Microbiology, 1097-1106, (2011)
- [7]. Francis Amala Rejula and Masilamai Dhinakaran, Removal of Zinc (II) by Non Living Biomass of *Agaricus Bisporus*, Research Journal of Recent Sciences, 1(9), 13- 17 (2012)
- [8]. Elisa Bayraktarov, Roy E. Price, Timothy G. Ferdelman and Kai Finster, The pH and pCO₂ dependence of sulfate reduction in shallow-sea hydrothermal CO₂ – venting sediments (Milos Island, Greece), Front. In Micro (2013)
- [9]. Mehar Fatma, M. Iqbal R., Khan Asim Massod and nafees A. Khan, Coordinate changes in assimilatory Sulphate reduction are correlated to salt tolerance : Involvement of Phytohormones, Annual review and Research in biology, 3(3) (2013)
- [10]. Z. Manafi 1, M. Hashemi 1, H. Abdollahi, Gregory. J. Olson, Bio-corrosion of water pipeline by sulphate reducing bacteria in a mining environment, Research Journal of Recent Sciences, 12(46), 6504-6516 (2013)
- [11]. Murhekar Gopalkrushna Haribhau, Trace Metals Contamination of Surface Water Samples in and Around Akot City in Maharashtra, India, Research Journal of Recent Sciences, 1(7), 5-9 (2012)
- [12]. Patil S. Z., Unnithan A. R. and Unnikrishnan G., Biosulphidogenesis and Bioaccumulation of sulphate by moderately Thermophilic, Facultative anaerobic Bacteria *Aeromonas hydrophila* isolated from hot Water spring, Research Journal of Recent Sciences, vol.3, 203-208 (2014).