

Bio-Cng as Transportation Fuel for Automobiles

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Abstract: Bio gas is the natural gas which is produced by anaerobic digestion of the organic mass or biodegradable matter. It can be used as a fuel in order to supply energy. This paper deals with the production of biogas from different sources and its conversion into BIO CNG which can be replaced with the current fossil fuels so that the automobiles can be energized, Simultaneously this also helps in the reduction of emissions of harmful gases into the environment which leads to sustainable living of the mankind.

Keywords: BIO-CNG, biogas, alternative fuels, compressed biogas.

I. Introduction

Due to the increased population and advancements made in the automotive sector the usage of the fossil fuels has been increased drastically all around the world which led to the pollution of the environment and also global warming. The emissions due to fossil fuels contains many harmful and greenhouse gases like CO, H₂S, NO_x, CO₂ etc. India's consumption of crude oil is approximately 3.2 million barrels per day India produces 1.4 metric tons of Carbon Dioxide per capita annually. World consumption of crude oil is approximately 86.9 million barrels per day. World reserves of crude oil are reported to be 687.43 billion barrels. At the present rate of consumption the reserves are expected to last for 25 more years [1]. So the increased concern regarding the fossil fuels has led to the research for an alternative fuel like biogas According to an Ag STAR report, a 1MW anaerobic digestion facility can produce approximately 3million kWh of electricity per year, which is enough to supply power to more than 200 homes [2]. Another study evaluated energy crops for biogas production in the EU-25 (the 25 Member States of the European Union).It showed that 320 million tons of crude oil equivalents(COE)could be produced with crop rotations that integrate the production of food ,feed , raw materials (e. g . oils ,fats, organic acids),which would provide up to 96% of the total energy demand of cars and trucks In the EU-25 [3]. In China, biogas production from small-scale biogas digester has increased from approximately1.8_109 m³ in 1996to1.0_1010 m³ in 2007 (equivalentto1.1_1011 kWh electricity), while biogas production from medium-and large-scale biogas projects has increased from approximately1.2_1011 m³ in 1996 to 6.0_1012 m³ in 2007 (equivalentto6.3_1013 kWh electricity)

Biogas is a chemical mixture of methane carbon dioxide nitrogen hydrogen sulphide and oxygen which are in the proportions as tabulated below [5]

Contents	% percentage
Methane	50-75
Carbon dioxide	25-50
Nitrogen	0-10
Hydrogen	0-1
Hydrogen Sulphide	0-3
Oxygen	0-0.5

II. Biogas Production

The bio gas production can be done in many ways the most popular among them are by

- 1) animal dung
- 2) Food waste
- 3) Municipality waste
- 4) Agricultural waste

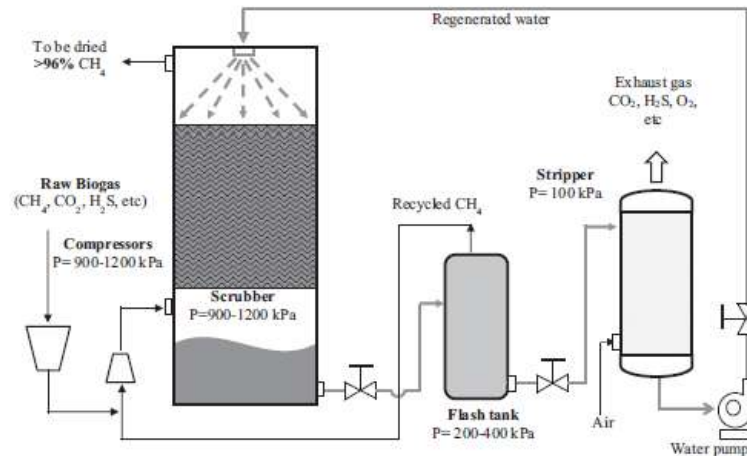
III. Cleaning Of Biogas

The biogas generated by the above mentioned process contains a lower hydrocarbons. If biogas is used for production of heat and electricity via heat and power (CHP) unit, only water and H₂S removal is required. However, using biogas transportation fuel conversion has strict requirements on its composition. So this process of increasing the methane content in the biogas is known as cleaning of biogas
Cleaning of biogas can be done by the following methods [6]

- 1) Pressurized Water scrubbing
- 2) Pressure swing adsorption
- 3) Amine adsorption
- 4) Membrane permeation

1) Pressurized water scrubbing:

Pressurized water scrubbing (PWS) is the most commonly used biogas cleaning method. This method takes advantage of the higher water solubility of CO_2 and H_2S compared to CH_4 , thereby separating both CO_2 and H_2S simultaneously from biogas with a high efficiency. A schematic representation of this method is shown in the below fig.

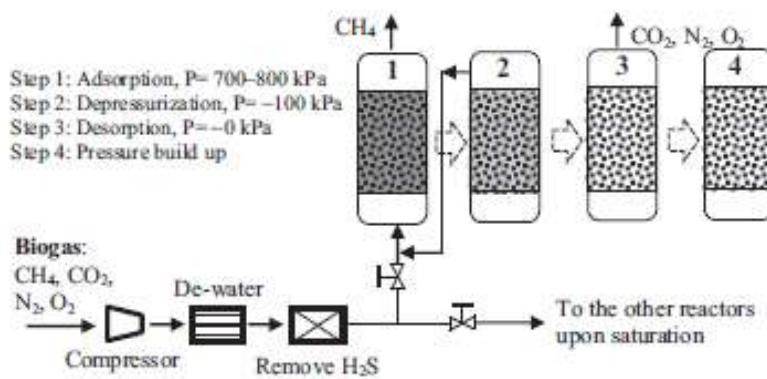


To increase the absorption of CO_2 and H_2S , biogas is usually compressed to 900–1200 kPa and a high surface area packing media is used. Inside the scrubber, biogas flows counter-currently (opposite direction) to water that is sprayed from the top of the scrubber, and the absorption primarily occurs on the surface of the packing media. Cleaned biogas can contain 96% CH_4 after drying.

The advantages of this method is it doesn't use any chemicals and it simultaneously removes both CO_2 and H_2S . Its disadvantage is that it uses a large amount of water. A recent study has proved that if the pressure content applied is increased then the water usage can be decreased.

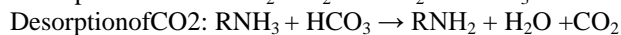
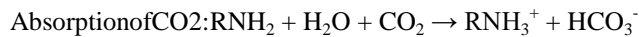
2) Pressure swing adsorption:

Pressure swing adsorption (PSA) uses the adsorbent's differences in gas adsorption rates to capture preferred gases (e.g. CO_2 , O_2 , and N_2) at a high pressure, and then releases the adsorbents at a low pressure to regenerate the adsorbent for a subsequent adsorption cycle. Commonly used adsorbents are zeolite, carbon molecular sieve, silica gel, and activated carbon, due to their low cost, large specific area and pore volume, and excellent thermal stability. These adsorbents are designed to have a specific pore size thus enabling selective adsorption of molecules that are smaller than the designed pore size. The molecular size of CH_4 , CO_2 , O_2 , and N_2 are 4.0, 2.8, 2.8, and 3.0 Å, respectively, at standard conditions. Therefore, an adsorbent with a pore size of 3.7 Å is able to capture CO_2 , O_2 , and N_2 , but not CH_4 , thereby cleaning the biogas. This method was first employed in 1989 for CH_4 enrichment from biogas when Pande and Fabiani used a natural zeolite used extensively for biogas cleaning. The PSA adsorption and desorption process usually includes four steps as represented in schematic diagram. In the pressurized vessel (700–800 kPa, step 1), CO_2 and other small-size gases are adsorbed, while the enriched CH_4 leaves from the top of the vessel. When the adsorbent is saturated by adsorbents, the biogas flows to another vessel. It usually needs four or more vessels operating at the same time to create a continuous operation, which reduces the energy needed for gas compression. When a vessel becomes saturated, it goes through a depressurization step in which the pressure is reduced to around atmospheric condition (100 kPa, step 2). Gas released in this step contains both impurities and methane, and is recycled through the desorption vessel. The pressure of the vessel is then further decreased to near vacuum (0 kPa, step 3), which de-adsorbs captured impurities, regenerating the adsorbents. The gas that leaves the vessel in this step mainly consists of CO_2 , N_2 and O_2 . The pressure is built up in (step 4) for a subsequent cleaning cycle.



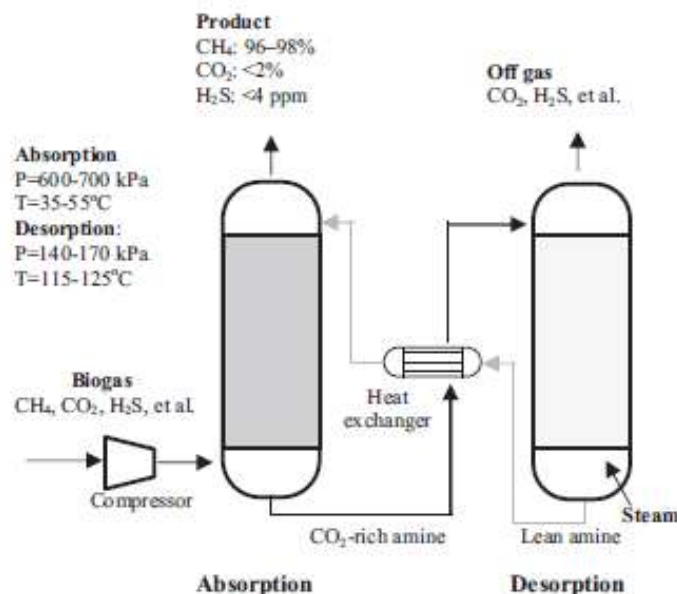
3) Amine adsorption:

Amine solvent has a high absorption selectivity of CO₂; therefore, is often used to separate CO₂ from gas streams. Amine absorption technology was originally developed for separating CO₂ from coal fired power plant flue gas in the early 1980s, and later was adopted as a biogas cleaning technology. Commonly used solvents are alkanolamines, such as monoethanolamine (MEA), diethanolamine (DEA) or methyldiethanolamine (MDEA), among which MEA is the most widely employed solvent for low pressure absorption. These solvents not only enhance CO₂ absorption capacity but also reduce corrosion problems. The reactions during adsorption and desorption processes are shown below.



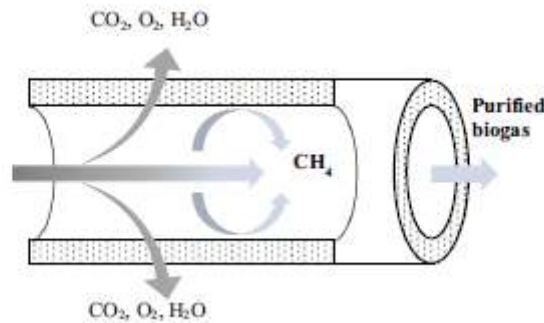
Where R is an organic component. For example, R is - (CH₂)₂OH for MEA. The above reactions are mainly governed by temperature and pressure. Low temperature and high pressure favor absorption,

while high temperature and low pressure promote desorption. A schematic diagram of amine absorption and desorption is shown below. Biogas is usually compressed at 600–700 kPa before feeding into the absorption reactor. In the absorption phase, CO₂ and some H₂S gas dissolve into the solvent, while high-purity CH₄ gas leaves the reactor. The CO₂-rich solvent is then transferred to the regeneration reactor. To accelerate desorption, high temperature (115–125°C) and relatively low pressure (140–170 kPa) are usually employed. After desorption, a high-purity CO₂ product can be collected and the CO₂ lean solvent is returned to the absorption reactor. The main disadvantage of this method is its high demand for amine.



4) Membrane permeation:

The design principle of membrane permeation is that under a certain pressure, gases with high permeability (e.g. small molecular size and low affinity) can be transported through the membrane while gases with low permeability are retained. High permeable impurities, such as CO₂, O₂, and H₂O, pass through the membrane as permeate, while low permeable CH₄ is retained and can be collected at the end of the hollow column. Using membranes for bio gas treatment began in the early 1980s when Kimura and Walmet made polymer membranes and used them to separate a synthetic mixture of CH₄ and CO₂. After decades of development, the membrane permeation method is now known for its safety, scale-up flexibility, simplicity of operation and maintenance, and no requirement for hazardous chemicals. General criteria for evaluating membrane separation are selectivity, pressure drop, CH₄ loss, and membrane life span. In 1983, a cellulose acetate spiral-wound membrane was used to treat biogas. During the 18 month trial, 96.5% CH₄ content was obtained.



IV. Bio-Cng and Its storage:

The BIO CNG differs from CNG in its chemical composition. Even though both the gases contain methane, carbon dioxide, Hydrogen sulphide, nitrogen, oxygen. CNG additionally contains ethane and propane including the above chemicals mentioned. Cleaned (greater than 97% CH₄ purity) biogas can be compressed to produce BIO-CNG, which is very similar to regular CNG. BIO-CNG is also known as compressed biomethane. Conversion of biogas into BIO-CNG requires removal of impurities using the four methods as mentioned above. Cleaned biogas should contain more than 97% CH₄ and less than 2% O₂. BIO-CNG is then produced by compressing (20–25 MPa) clean biogas to less than 1 percent of the volume it occupies at standard atmospheric pressure. A typical BIO-CNG station usually is composed of an impurity separation unit, a multi-stage compressor, and a high pressure storage tank. There are two commonly applied storage systems in industry

- 1) Buffer storage
- 2) Cascade storage.

The buffer storage system maintains the pressure of CNG in the range of 20.5–25 MPa, and provides CNG with a maximum pressure of 20 MPa to a vehicle's on-board cylinders. In this case, all filling station reservoirs are connected and maintained at the same pressure. The cascade storage system is typically composed of three reservoirs with low, medium, and high pressure, respectively, and filling CNG on-board cylinders takes three steps. The vehicle's on-board cylinders are firstly connected to the low-pressure reservoir. As the pressure in the reservoir declines and that in the on-board cylinder increases, the gas flow rate decreases. When the flow rate has declined to a pre-set level, the system switches to the medium-pressure reservoir, then finally to the high-pressure reservoir to complete the filling. Oppositely, when refilling the reservoirs, the high-pressure reservoir is prioritized, and then followed by the medium and low reservoirs. This method ensures that the high-pressure reservoir (used to complete the fill) is maintained at a maximum pressure all times, ensuring that vehicles are always supplied with the maximum amount of gas. Compared to the buffer storage system, the cascade system consumes about 50% less energy but charges 20% less biogas and takes three times longer to fill. Therefore, the cascade system is preferred for filling fleet vehicles that usually takes hours (time-fill), while the buffer system meets the needs for fast-fill that can be completed within five minutes.

V. Advantages Ofbio-Cng:

1) Emissions:one of the major advantages of biogas is its low greenhouse gases emissions. GHG emissions can be reduced up to 90% with the help of BIO-CNG

When a natural gas driven vehicle is made to run by replacing the fuel with BIO-CNG the following reduction inthe emissions of the GHG (greenhouse gases) has been observed [7]:

Greenhouse gases	% reduction in emissions
1) Carbon monoxide	70 -90
2) Non methane organic gas	50-70
3) Nitrogen oxides	75-95
4) Carbon dioxide	20-30

2) Calorific value: BIO-CNG which is obtained by cleaning of biogas has a high calorific value when compared to other fuels and it also meets SAE J1616 standards for CNG [8]

Fuel	Calorific value in KJ/Kg
CNG	53000*
Petrol	48000*
Diesel	44000*
LPG	49789*
BIO CNG	52000*

* Slightly differs according to their hydrocarbons content

3) Cost:The cost required for the production of BIO-CNG is approximately 50% of that of petrol and diesel

Fuel	Cost (INR) per liter [9]
Petrol	62/-
Diesel	56/-
LPG	64.5/-
CNG	37.2/-
BIO CNG	35*/-[10]

*estimated cost

VI. Conclusion

Hence by the above benefits which are mentioned BIOCNG can be considered as the perfect replacement for the current fossil fuels like petrol and diesel. more over in a country like India which is one of the most polluted country after China,USAand European nations [11]there is an urgent need for this kind of alternative fuel replacement recently a BIOCNG plant has been established in Mahindra city Chennai in IndiaThe plant will convert 8 tons of food and kitchen waste generated every day at the MWC into 1,000 cubic meter of raw biogas. Further, the raw bio gas can yield 400 kg per day of purified CNG-grade fuel which is equivalent to a 200 kW power plant. As a by-product, four tons of organic fertilizer will be produced per day[12].There is more need for this kind of development not only in India but all around the world these kind of activities help in sustainable living of the mankind.

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